A Convenient Synthetic Route to Mannose 6-Phosphonate–Cholesteryl Conjugate

Sébastien Vidal, Alain Morère, and Jean-Louis Montero

Laboratoire de Chimie Biomoléculaire (UMR 5032), Université Montpellier II, ENSCM, 8 Rue de l'Ecole Normale, 34296—Montpellier Cedex 05, France

Received 26 September 2002

ABSTRACT: A multistep synthesis of a mannose 6phosphonate-based glycolipid is described involving (1) a one-carbon chain elongation at the 6-position of mannose, followed by (2) phosphonation, using tris(trimethylsilyl)phosphite. This method was shown to be efficient and provides a general route to various mannose 6-phosphonate-based compounds for the design of drug delivery systems. © 2003 Wiley Periodicals, Inc. Heteroatom Chem 14:241–246, 2003; Published online in Wiley InterScience (www.interscience.wiley.com). DOI 10.1002/hc.10134

INTRODUCTION

Lectins are involved in a wide variety of biological functions such as cell–cell communication, virus– cell, toxin–cell, or bacterium–cell interactions, all of which occur through lectin–carbohydrate recognition at the cell surface [1]. As part of our ongoing work, we focus on one lectin target from the P-type family, namely, the Mannose 6-Phosphate/Insulinlike II Receptor (M6P/IGFIIR). This 300 kDa type I transmembrane glycoprotein was shown [2] to bind Mannose 6-Phosphate (M6P)-bearing glycoproteins with high affinity (µM–nM range) and then internalize them through endocytosis of the receptor–ligand

© 2003 Wiley Periodicals, Inc.

complex [3]. We recently [4] demonstrated that an amphiphilic steroidal mannose 6-phosphonate can be incorporated into the liposome lipidic bilayer and that the resulting "functionalized" vesicle interacted strongly with MCF-7 breast cancer cells (overexpressing the M6P/IGFIIR), demonstrating the high potential of such system for drug delivery applications.

However, during the preparation of M6Pn/CSA [4] (where M6Pn and CSA stands for mannose 6-phosphonate and CholesterylSuccinylAnilinyl, respectively) used in that investigation, we observed (Fig. 1) partial degradation of this molecule, using Rabinowitz's procedure [5] for the deprotection of the diethylphosphonate (using Me₃SiBr), and partial cleavage of the anomeric bond and/or amide bond(s).

It is worth pointing out that similar degradation problems have been described previously for aminoacids [6] or carbohydrate [7] frameworks. We therefore decided to prepare a modified amphiphilic M6Pn compound in which the hydrophilic carbohydrate head will be linked to the hydrophobic steroidal tail through a triethyleneglycol (TEG) moiety (Fig. 1).

RESULTS AND DISCUSSION

We decided to use the Arbuzov phosphonation as the key step to introduce the phosphonate moiety at the very end of our synthetic scheme. This approach allowed us to use *tris*(trimethylsilyl)phosphite, which reacts with an alkyl halide derivative affording *bis*-silylated phosphonates, which are easily hydrolyzed

Correspondence to: Alain Morère; e-mail: morere@univ-montp2.fr.

Contract grant sponsor: Scientific Council of the Université Montpellier II.

Contract grant sponsor: Fondation pour la Recherche Médicale. Contract grant sponsor: Mayoly-Spindler Laboratories.

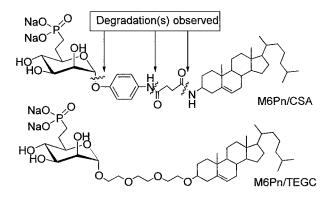


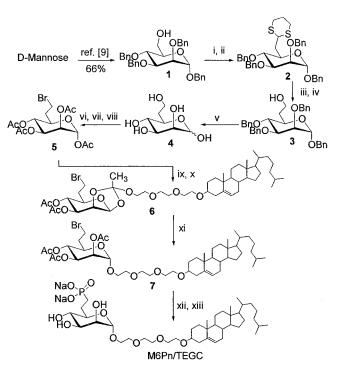
FIGURE 1 Partial degradations of M6Pn/CSA observed and structure of the target glycolipid M6Pn/TEGC.

to the corresponding phosphonic acids through mild basic hydrolysis with saturated $NaHCO_{3(aq)}$ during the reaction work-up.

This new strategy requires a one-carbon chain elongation at the 6-position of mannose to obtain the alkyl halide intermediate. This mannose derivative was therefore prepared through (1) a cautiously selected one-carbon chain elongation method and (2) a glycosylation reaction.

The synthesis of the M6Pn/TEGC was accomplished as outlined in Scheme 1. Many studies have been performed concerning the one-carbon chain elongation of various alcohols [8]. From these methods, we decided to use a one-carbon chain elongation at the 6-position of benzyl 2,3,4-tri-*O*-benzyl- α -D-mannopyranoside [9] (1) through triflate activation of the primary alcohol followed by substitution [10] with 2-lithio-1,3-dithiane affording compound **2** in 80% yield over both steps.

The dithioacetal 2 was subsequently unmasked to the aldehyde [11] (not isolated), which was reduced to furnish the homologated mannoheptose derivative 3 in 90% yield over both steps. This fourstep procedure could be performed rapidly since each reaction in the synthetic scheme only takes 5-10 min and only one purification step is necessary, i.e., purification of the 1,3-dithianyl derivative 2. Hydrogenolysis of 3, using a catalytic amount of acetic acid [12] to aid in the deprotection of the anomeric center, afforded the deprotected 6-deoxy-Dmannoheptose 4. Next, we had to functionalize the pyranose ring to allow (1) the introduction of the steroidal moiety via glycosylation and (2) the phosphonation reaction to obtain the final amphiphilic phosphonate. Thus, compound 4 was first activated at the 7-position, using tosyl chloride in pyridine followed by *in situ* peracetylation of the pyranose ring [13] and subsequent bromination utilizing lithium



SCHEME 1 Preparation of the M6Pn/TEGC. Reagents and conditions: (i) Tf₂O, CH₂Cl₂, 2,6-di-*t*-butyl-4-methylpyridine, -10°C to r.t., 5 min; (ii) 1,3-dithiane, THF, n-BuLi, HMPA, -78°C to r.t., 10 min, 80% over both steps; (iii) CH₃I, CH_3CN/H_2O (11:2), $CaCO_3$, $50^{\circ}C$, 3 h; (iv) $NaBH_4$, EtOH/H2O (1:1), r.t., 10 min, 90% over both steps; (v) H2, THF/H2O/AcOH (50:50:1), Pd(OH)2/C (20%), r.t., 24 h, 98%; (vi) TsCl, C₅H₅N, r.t., 5 h; (vii) Ac₂O, C₅H₅N, 0°C to r.t., 12 h; (viii) LiBr, butanone, 85°C, 1 h, 45% over three steps; (ix) HBr/AcOH, CH₂Cl₂, r.t., 16 h; (x) 8-(cholest-5-en-3\beta-yloxy)-3,6-dioxaoctan-1-ol, CH₂Cl₂, AgOTf, sym-collidine, 0°C to r.t., 15 min, 45% over both steps; (xi) TMSOTf, CH₂Cl₂, molecular sieves (3 Å), -5°C, 5 min, 40%; (xii) P(OSiMe₃)₃, 60°C, 16 h then NaHCO3(aq); (xiii) NaOMe, MeOH, r.t., 16 h, 55% over both steps.

bromide in refluxing butanone [14], affording compound **5** in 45% yield over three steps. The poor yield obtained for this one-pot synthesis is probably a consequence of the low solubility of compound **4** in pyridine. Koenigs–Knorr glycosylation [15] of **5**, using known 8-(cholest-5-en-3 β -yloxy)-3,6-dioxaoctan-1-ol [16] as the glycosyl acceptor, affords the orthoester derivative **6** as the major product, which upon subsequent treatment with a Lewis acid underwent rearrangement [17] to furnish the desired glycoside **7**.

The final step in our synthesis involved an Arbuzov reaction between the alkyl bromide derivative **7** and *tris*(trimethylsilyl)phosphite, followed by deprotection of the acetate under Zemplén conditions affording the final M6Pn/TEGC in good yields.

CONCLUSION

In summary, we have demonstrated that a new route to steroidal M6Pn is possible through (1) one-carbon chain elongation at the 6-position of mannose, followed by (2) phosphonation using the tris(trimethylsilyl)phosphite under Arbuzov conditions. Preparation of liposomes incorporating M6Pn/TEGC and evaluation of their interactions with MCF-7 cells are now being investigated in our group with the aim of developing a new drug delivery system. Although the overall yield is still low (1.9% over 18 steps starting from *D*-mannose), this strategy provides a general route to a wide variety of functionalized M6Pn where the anomeric center will be substituted with any biologically active molecule during the glycosylation.

EXPERIMENTAL

General Methods

Analytical TLC were performed using aluminumcoated TLC plates 60-F₂₅₄ (Merck). Plates were developed with (1) UV light (254 nm), and (2) immersion in a 10% H₂SO₄/EtOH solution followed by charring or (3) immersion in a 5% Rhodanine/EtOH solution followed by charring (for aldehydes), or (4) immersion in a phosphomolybdic solution (for phosphorus containing compounds). Silica gel column chromatography was performed with silica gel 60A (Carlo Erba). Optical rotations were measured at the sodium D-line with a Perkin-Elmer-241 polarimeter. IR Spectra were obtained on Perkin-Elmer FT-1600 spectrometer in CH₂Cl₂ solutions. Fast Atom Bombardment (FAB) mass spectra were recorded on a Jeol JMS-DX300 spectrometer in either positive (>0) or negative (<0) modes and using either 3nitrobenzylic alcohol (NBA) or glycerol/thioglycerol (1:1) mixture (G/T). ¹H NMR Spectra were recorded on a Brüker DRX 400 (400 MHz), at 25°C. Chemical shifts (δ) are given in ppm and referenced using residual solvent signals (7.24 ppm for CHCl₃ and 4.79 ppm for HOD). The following abbreviations were used to explain the signal multiplicities or characteristics: s (singlet), d (doublet), dd (double doublet), t (triplet), td (triplet doublet), m (multiplet). ¹³C NMR Spectra were recorded on a Brüker DRX 400 (100.6 MHz). Chemical shifts (δ) are given in ppm relative to TMS as an external reference. ³¹P NMR Spectra were recorded on a Brüker DPX 200 (81.0 MHz). Chemical shifts (δ) are given in ppm relative to phosphoric acid (85%) as an external reference.

Benzyl 6-Deoxy-6-(2'-dithianyl)-2,3,4-tri-Obenzyl- α -D-mannopyranoside (**2**)

To a cooled $(-10^{\circ}C)$ solution of benzyl 2,3,4-tri-Obenzyl-6-deoxy- α -D-mannopyranoside (1) [9] (3.29 g, 6.08 mmol) and 2,6-di-tert-butyl-4-methylpyridine (1.56 g, 7.6 mmol) in CH₂Cl₂ (20 ml) was added triflic anhydride (1.79 ml, 6.7 mmol). The reaction mixture was stirred at r.t. for 10 min then neutralized with NaHCO_{3(aq)} (200 ml, 0.1 g/l) and the aqueous layer was extracted with CH_2Cl_2 (3 × 300 ml). Organic layers were combined, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The crude triflate was used for next step without further purification. To a cooled $(-78^{\circ}C)$ solution of 1,3-dithiane (2.56 g, 21.29 mmol) in THF (20 ml) was added HMPA (1.06 ml, 6.08 mmol). Next, n-BuLi (5 ml, 21.29 mmol, 1.6 M in hexanes) was added dropwise. After 10 min, a solution of triflate dissolved in THF (30 ml) was added dropwise, and after a further 10 min, the reaction was neutralized with NH₄Cl_(aq) (200 ml, 0.1 g/l) and the aqueous layer extracted with CH_2Cl_2 (2 × 300 ml). Organic layers were combined, dried (Na_2SO_4) , filtered, and evaporated under reduced pressure. The residue was purified by silica gel column chromatography (light petroleum/ether 4:1) affording 2 (2.87 g, 80%). $R_f = 0.50$ (light petroleum/ether 7:3). $[\alpha]^{20}_{\rm D} + 31.0^{\circ}$ (c 0.72, CHCl₃). ¹H NMR (CDCl₃): $\delta = 1.80-2.20$ (m, 3H, H-6a and CH₂CH₂S), 2.35–2.55 (m, 1H, H-6b), $2.65-2.90 \text{ (m, 4H, CH}_2\text{C}H_2\text{S}), 3.76 \text{ (t, 1H, } J = 9.4 \text{ Hz},$ H-4), 3.86 (dd, 1H, J = 1.9, 3.0 Hz, H-2), 4.00 (dd, 1H, H-3), 4.10 (td, 1H, J = 2.3, 9.4 Hz, H-5), 4.38 (dd, 1H, J = 4.0, 11.1 Hz, H-7), 4.45 (d, 1H, J = 11.5 Hz, CH₂Ph_{anom}), 4.65 (s, 2H, CH₂Ph), 4.69 (d, 1H, J = 10.9 Hz, CH_2 Ph), 4.73 (d, 1H, J = 12.4 Hz, CH₂Ph), 4.81 (d, 1H, CH₂Ph), 4.90 (d, 1H, CH₂Ph_{anom}), 4.92 (d, 1H, H-1), 5.00 (d, 1H, CH₂Ph), 7.30–7.45 (m, 20H, H-ar). ¹³C NMR (CDCl₃): $\delta = 26.5$ (CH₂CH₂S), 29.7, 30.3 (CH₂CH₂S), 38.2 (C-6), 43.8 (C-7), 68.6-80.7 (C-2, C-3, C-4, and C-5), 69.4 (CH₂Ph_{anom}), 72.7, 73.3, 75.9 (CH₂Ph), 97.4 (C-1), 128.0–129.4 (CHar), 137–137.5 (C_{quat}-ar). MS FAB > 0 (NBA): m/z =589 (M)⁺, 551 (M-CH₂Ph)⁺, 535 (M-OCH₂Ph)⁺, 443 (M-OCH₂Ph-CH₂Ph)⁺, 91 (CH₂Ph)⁺. Anal for C₃₈H₄₂O₅S₂: Calcd: C 71.00, H 6.59; Found: C 71.28, H 6.64.

Benzyl 6-Deoxy-2,3, 4-tri-O-benzyl- α -D-mannoheptopyranoside (**3**)

To a solution of **2** (1 g, 1.56 mmol) dissolved in CH_3CN/H_2O (52 ml, 11:2) was added calcium carbonate (470 mg, 4.67 mmol) and methyl iodide (2.15 ml, 34.3 mmol). The reaction was heated (50°C) for 3 h then neutralized with saturated $Na_2S_2O_{3(aq)}$ (100 ml). The aqueous layer was extracted with EtOAc $(2 \times 100 \text{ ml})$, and the organic layers combined, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The crude product was dissolved in absolute ethanol (25 ml) and a solution of sodium borohydride (148 mg, 3.89 mmol), dissolved in EtOH/H₂O (25 ml, 1:1) was added dropwise. The reaction was neutralized with 1 M HCl to $pH \approx 6$. The reaction mixture was then diluted with H_2O (150 ml), and the aqueous layer extracted with EtOAc (5×100 ml). The organic layers were combined, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The residue was purified by silica gel column chromatography (light petroleum/ether 3:7) affording 3 (778 mg, 90%). $R_f = 0.44$ (light petroleum/ether 3:7). $[\alpha]^{20}_{\rm D} +$ 54.9° (c 1.02, CHCl₃). IR (NaCl): $\nu = 3460$. ¹H NMR (CDCl₃): $\delta = 1.80-2.00$ (m, 1H, H-6a), 2.10-2.30 (m, 1H, H-6b), 3.75-4.05 (m, 6H, H-2, H-3, H-4, H-5, H-7a, and H-7b), 4.46 (d, 1H, J =11.8 Hz, CH₂Ph), 4.66 (s, 2H, CH₂Ph), 4.67 (d, 1H, J = 10.8 Hz, CH_2Ph_{anom}), 4.71 (d, 1H, J = 12.4 Hz, CH₂Ph), 4.72 (d, 1H, CH₂Ph), 4.80 (d, 1H, CH₂Ph) $4.89 (d, 1H, J = 1.6 Hz, H-1), 5.01 (d, 1H, CH_2Ph_{anom}),$ 7.20–7.45 (m, 20H, H-ar). ¹³C NMR (CDCl₃): $\delta =$ 34.2 (C-6), 61.5 (C-7), 69.4 (CH₂Ph_{anom}), 72.2-80.6 (C-2, C-3, C-4, and C-5), 72.6, 73.3, 75.8 (CH₂Ph), 97.5 (C-1), 128.0–128.9 (CH-ar), 137.5–138.8 (Couatar). MS FAB > 0 (NBA): $m/z = 577 (M + Na)^+$, 553 (M-H)⁺, 447 (M-OCH₂Ph)⁺, 91 (CH₂Ph)⁺. Anal for C₃₅H₃₈O₆: Calcd: C 75.79, H 6.91; Found: C 75.96, H 6.95.

6-Deoxy- α -D-manno-heptopyranose (4)

A solution of **3** (4.44 g) and Pd(OH)₂/C (20%, 2.5 g) in THF/H₂O (100 ml, 1:1) and glacial acetic acid (5 ml) was vigorously stirred under a hydrogen atmosphere for 24 h. The reaction was then filtered through celite and the filtrate concentrated under reduced pressure then freeze-dried from H₂O affording **4** (1.52 g, 98%). $R_f = 0.50$ (CH₂Cl₂/MeOH 7:3). ¹H NMR (D₂O): $\delta = 1.40-2.10$ (m, 2H, H-6a and H-6b), 3.20-3.85 (m, 6H, H-2, H-3, H-4, H-5, H-7a, and H-7b), 4.70-4.75 (m, 1H, H-1B), 4.98-5.02 (m, 1H, H-1 α). ¹³C NMR (D₂O): δ = 33.5, 33.6 (C-6α and C-6β), 58.8, 58.6 (C-7α and C-7β), 69.4–73.6 (C-2α, C-2β, C-3α, C-3β, C-4α, C-4β, C-5α, and C-5β), 94.0, 94.3 (C-1α and C-1β). MS FAB > 0 (G/T): $m/z = 195 (M + H)^+$, 177 (M-OH)⁺. Anal for C₇H₁₄O₆: Calcd: C 43.30, H 7.27; Found: C 43.13, H 7.32.

7-Bromo-6,7-dideoxy-1,2,3,4-tri-O-acetyl- α -Dmanno-heptopyranoside (**5**)

To a solution of 4 (730 mg, 3.76 mmol) in C_5H_5N (50 ml) was added tosyl chloride (1.08 g, 5.64 mmol). Reaction was stirred for 5 h, then Ac_2O (7.2 ml, 75.26 mmol) was added to the cooled $(0^{\circ}C)$ reaction mixture. After 12 h, the reaction was diluted with brine (125 ml) and the aqueous layer extracted with CH_2Cl_2 (3 × 150 ml). The organic layers were combined, dried (Na₂SO₄), filtered, and evaporated under reduced pressure, utilizing toluene $(3 \times 100 \text{ ml})$ to coevaporate C₅H₅N. The crude product was dissolved in butanone (20 ml) and lithium bromide (1 g, 11.5 mmol) added. The reaction was heated (85°C) for 1 h then diluted with brine (100 ml). The aqueous layer was extracted with CH_2Cl_2 (3 × 100 ml) and the organic layers were combined, dried (Na_2SO_4) , filtered, and evaporated under reduced pressure. The resulting residue was purified by silica gel column chromatography (light petroleum/ether 9:1) affording **5** (716 mg, 45%). $R_f = 0.61$ (light petroleum/ether 3:7) ¹H NMR (CDCl₃): $\delta = 1.80-2.00$ (m, 2H, H-6a and H-6b), 2.00 (s, 3H, CH₃CO), 2.07 (s, 3H, CH₃CO), 2.15 (s, 3H, CH₃CO), 3.40–3.70 (m, 2H, H-7a and H-7b), 4.08 (td, 1H, J = 3.0, 9.9 Hz, H-5), 5.16 (t, 1H, H-4), 5.26 (dd, 1H, J = 1.8, 3.5 Hz, H-2), 5.34 (dd, 1H, H-3), 5.99 (d, 1H, H-1). ¹³C NMR (CDCl₃): $\delta = 20.9-21.2$ (CH₃CO), 29.2 (C-7), 33.9 (C-6), 68.4–69.6 (C-2, C-3, C-4, and C-5), 90.6 (C-1), 170.1, 170.2, 170.3, 170.4 (CH₃CO). MS FAB > 0 (NBA): $m/z = 447 (M + Na)^+$, 365 (M-OAc)⁺, 345 (M-Br)+, 245 (M-2OAc-2H)+, 203 (M-2OAc- $CH_3CO-H)^+$, 43 (CH_3CO)⁺. Anal for $C_{15}H_{21}BrO_9$: Calcd: C 42.37, H 4.95; Found: C 42.29, H 4.92.

7-Bromo-6, 7-dideoxy-2, 3, 4-tri-O-acetyl-1, 2-O-[8-(cholest-5-en-3 β -yloxy)-3, 6-dioxaoctan-1yloxyethylidene]- β -D-manno-heptopyranose (**6**)

To a solution of **5** (220 mg, 0.52 mmol) in CH₂Cl₂ (15 ml) was added HBr (2.28 ml, 13 mmol, 5.7 M in acetic acid). The reaction was stirred at r.t. for 16 h and then neutralized with saturated NaHCO_{3(aq)} (50 ml). The aqueous layer was extracted with CH₂Cl₂ (3×70 ml), and the organic layers combined, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The crude product dissolved in CH₂Cl₂ (15 ml) was added dropwise to a solution of 8-(cholest-5en-3β-yloxy)-3,6-dioxaoctan-1-ol (287 mg, 0.57 mmol), silver trifluoromethanesulfonate (146 mg, 0.57 mmol) and *sym*-collidine (76 µl, 0.57 mmol) in CH₂Cl₂ (5 ml). The reaction was stirred at RT for 15 min before filtering through celite. The filtrate was concentrated under reduced pressure,

and the residue purified by silica gel column chromatography (light petroleum/ether 4:1 \rightarrow ether) affording **6** (210 mg, 45%). $R_f = 0.41$ (light petroleum/ether 1:4). ¹H NMR (CDCl₃): $\delta = 0.67$ (s, 3H, H-18'), 0.86 (d, 6H, J = 6.6 Hz, H-26'), 0.91 (d, 3H, J = 6.4 Hz, H-21'), 0.99 (s, 3H, H-19'), 0.80-2.50 (m, 30H, H-Chol H-6a and H-6b), 1.71 (s, 3H, Orthoester-CH₃), 2.07 (s, 3H, CH₃CO), 2.11 (s, 3H, CH₃CO), 3.10–3.30 (m, 1H, H-3'), 3.45–3.70 (m, 15H, OCH₂CH₂O, H-5, H-7a, and H-7b), 4.59 (dd, 1H, J = 2.6, 3.5 Hz, H-2), 5.00–5.20 (m, 2H, H-3 and H-4), 5.30-5.38 (m, 1H, H-6'), 5.44 (d, 1H, H-1). ¹³C NMR (CDCl₃): $\delta = 12.2$ (C-18'), 15.6– 41.7 (C-Chol, Orthoester-CH₃ and C-6), 21.1, 21.2 (CH₃CO), 29.1 (C-7), 50.6, 56.5, 57.2 (C-9', C-14', and C-17'), 62.4, 66.2 (OCH₂CH₂O), 67.7 (C-4), 69.0, 70.3 (OCH₂CH₂O), 70.9 (C-3), 71.0 (OCH₂CH₂O), 71.1 (C-5), 71.3 (OCH₂CH₂O), 76.8 (C-2), 79.9 (C-3'), 97.9 (C-1), 121.9 (Orthoester-*C*_{quat}), 141.4 (C-5'), 170.2, 170.7 $(CH_3CO).$

8-(Cholest-5-en- 3β -yloxy)-3,6-dioxaoctan-1-yl-7-Bromo-6,7-dideoxy-2,3,4-tri-O-acetyl- α -Dmanno-heptopyranoside (**7**)

To a cooled $(-5^{\circ}C)$ solution of **6** (160 mg, 0.18 mmol) and molecular sieves (3 Å) in CH_2Cl_2 (5 ml) was added trimethylsilyl trifluoromethanesulfonate (3.3 μ l, 18 μ mol). The reaction was stirred for 5 min then neutralized with Et_3N (50 µl) before being diluted with H_2O (50 ml). The aqueous layer was extracted with CH_2Cl_2 (3 × 50 ml), and the organic layers combined, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The resulting residue was purified by silica gel column chromatography (light petroleum/ether 9:1 \rightarrow ether) affording 7 (65 mg, 40%). $R_f = 0.58$ (light petroleum/ether 1:4). ¹H NMR (CDCl₃): $\delta = 0.60$ (s, 3H, H-18'), 0.79 (d, 3H, J =6.6 Hz, H-26'), 0.80 (d, 3H, H-26'), 0.84 (d, 3H, J = 6.5 Hz, H-21'), 0.92 (s, 3H, H-19'), 0.70–2.35 (m, 28H, H-Chol), 1.85-1.95 (m, 1H, H-6a), 1.91 (s, 3H, CH₃CO), 1.99 (s, 3H, CH₃CO), 2.07 (s, 3H, CH₃CO), 2.00–2.10 (m, 1H, H-6b), 3.05–3.15 (m, 1H, H-3'), 3.40-3.50 (m, 1H, H-7a), 3.50-3.70 (m, 13H, OCH_2CH_2O and H-7b), 3.97 (td, 1H, J = 2.5, 9.9 Hz, H-5), 4.75 (d, 1H, J = 1.5 Hz, H-1), 5.04 (t, 1H, H-4), 5.21 (dd, 1H, J = 3.5 Hz, H-2), 5.25–5.30 (m, 1H, H-6'), 5.27 (dd, 1H, H-3). ¹³C NMR (CDCl₃): $\delta = 12.3$ (C-18'), 19.1–42.7 (C-Chol), 21.1, 21.2, 21.5 (CH₃CO), 29.2 (C-7), 34.9 (C-6), 50.6, 56.6, 57.2 (C-9', C-14', and C-17'), 62.6, 67.7 (OCH₂CH₂O), 67.9 (C-5), 69.5 (C-3), 69.8 (C-4), 70.1 (C-2), 70.5, 71.0, 71.2, 71.3 (OCH₂CH₂O), 79.9 (C-3'), 97.8 (C-1), 121.9 (C-6'), 141.4 (C-5'), 170.2, 170.3, 170.5 (CH₃CO). Anal for C₄₆H₇₅BrO₁₁: Calcd: C 62.50, H 8.55; Found: C 62.77, H 8.63.

8-(Cholest-5-en- 3β -yloxy)-3,6-dioxaoctan-1-yl-6-Deoxy-6-dihydroxyphosphinylmethylene- α -Dmannopyranoside Disodium Salt (M6Pn/TEGC)

A solution of 7 (67 mg, 76 µmol) in tris-(trimethylsilyl)phosphite (5 ml) was refluxed (160°C) for 16 h. The reaction mixture was poured into saturated NaHCO_{3(aq)} (50 ml). The aqueous layer</sub> was extracted with CH_2Cl_2 (3 × 50 ml), and the organic layers were combined, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The crude mixture (showing a single signal by ³¹P NMR spectroscopy at 27 ppm) was dissolved in MeOH (5 ml) and NaOMe (35 mg, 610 µmol) was added. After 24 h, the reaction was neutralized with 1 M HCl and concentrated under reduced pressure. The residue was purified by silica gel column chromatography (*i*-PrOH/NH₄OH/H₂O 6:3:1), then treated with ion exchange resin (DOWEX 50WX2, Na⁺ form) affording M6Pn/TEGC (31 mg, 55%). $R_f = 0.52$ (*i*-PrOH/NH₄OH/H₂O 6:3:1). ¹H NMR (D₂O): $\delta = 0.70-$ 2.30 (m, 47H, H-Chol, H-6a, H-6b, H-7a, and H-7b), 3.20–4.00 (m, 17H, OCH₂CH₂O, H-2, H-3, H-4, H-5, and H-3'), 5.30-5.40 (m, 1H, H-6'). ³¹P NMR (D₂O): $\delta = 23.0$ (s, $P(O)(ONa)_2$). MS FAB > O (G/T): $m/z = 803 (M + H)^+$, 781 (M-Na + 2H)⁺. MS FAB < 0 $(G/T): m/z = 757 (M-2Na + H)^{-}.$

REFERENCES

- [1] (a) Sharon, N.; Lis, H. Sci Am 1993, 268, 82–88; (b)
 Varki, A. Glycobiology 1993, 3, 97–130; (c) Lis, H.;
 Sharon, N. Chem Rev 1998, 98, 637–674.
- [2] (a) Tong, P. Y.; Gregoy, W.; Kornfeld, S. J Biol Chem 1989, 264, 7962–7969; (b) Tong, P. Y.; Kornfeld, S. J Biol Chem 1989, 264, 7970–7975.
- [3] (a) Dahms, N. M.; Lobel, P.; Kornfeld, S. J Biol Chem 1989, 264, 12115–12118; (b) Méresse, S.; Bauer, U.; Ludwig, T.; Mauxion, F.; Schmidt, A.; Hoflack, B. Med Sci 1993, 9, 148–156.
- [4] Barragan, V.; Menger, F. M.; Caran, K. L.; Vidil, C.; Morère, A.; Montero, J.-L. Chem Commun 2001, 85– 86.
- [5] Rabinowitz, R. J Org Chem 1963, 28, 2975–2978.
- [6] Garbay-Jaureguiberry, C.; Ficheux, D.; Roques, B. P. Int J Peptide Protein Res 1992, 39, 523–527.
- [7] (a) Le Maréchal, P.; Froussios, C.; Level, M.; Azerad, R. Carbohydr Res 1981, 94, 1–10; (b) Wang, M. F.; Crilley, M. M. L.; Golding, B. T.; McInally, T.; Robinson, D. H.; Tinker, A. J Chem Soc Commun 1991, 667–668.
- [8] For Several examples of one-carbon chain elongation of alcohols see: (a) Baer, H. H.; Hanna, H. R. Carbohydr Res 1982, 102, 169–183; (b) Rouzaud, D.; Sinaÿ, P. J Chem Soc Chem Commun 1983, 1353– 1354; (c) Burford, C.; Cooke, F.; Roy, G.; Magnus, P.

Tetrahedron 1983, 39, 867-876; (d) Anderson, R. Synthesis 1985, 717-734; (e) Jones, K.; Mood, W. W. Carbohydr Res 1986, 155, 217-222; (f) Baer, H. H.; Breton, L. R.; Shen, Y. Carbohydr Res 1990, 200, 377-398; (g) Goekjian, P. G.; Wu, T.-C.; Kang, H.-Y.; Kishi, Y. J Org Chem 1991, 56, 6422-6434; (h) van der Klein, P. A. M.; van Boom, J. H. Carbohydr Res 1992, 224, 193-200; (i) Baer, H. H.; Shen, Y.; Wu, X. Carbohydr Res 1993, 241, 117-129; (j) De Raddt, A.; Griengl, H.; Klempier, N. J Org Chem 1993, 58, 3179-3184; (k) Aspinall, G. O.; McDonald, A. G.; Sood, R. K. Can J Chem 1994, 72, 247-251; (l) Pakulski, Z.; Zamojski, A. Polish J Chem 1994, 68, 1109-1114; (m) El Garrouj, D.; Aliau, S.; Aumelas, A.; Borgna, J.-L. J Med Chem 1995, 38, 2339-2348; (n) Pakulski, Z.; Zamojski, A. Tetrahedron 1995, 51, 871-908; (o) Barbud, C.; Bols, M.; Lundt, I.; Sierks, M. R. Tetrahedron 1995, 51, 9063–9078; (p) Leeuwenburgh, M. A.; Picasso, S.; Overkleeft, H. S.; van der Marel, G. A.; Vogel, P.; van Boom, J. H. Eur J Org Chem 1999, 1185–1189; (q) Xin, Y.-C.; Zhang, Y.-M.; Mallet, J.-M.; Glaudemans, C. P. J.; Sinaÿ, P. Eur J Org Chem 1999, 471-476.

- [9] Dziewiszek, K.; Zamojski, A. Carbohydr Res 1986, 150, 163–171.
- [10] (a) Seebach, D.; Corey, E. J. J Org Chem 1975, 40, 231–237; (b) Shen, Q.; Sloss, D. G.; Berkowitz, D. B.

Synth Commun 1994, 24, 1519–1530; (c) Berkowitz, D. B.; Bose, M.; Pfannenstiel, T. J.; Doukov, T. J Org Chem 2000, 65, 4498–4508.

- [11] Greene, T. W.; Wuts, P. G. Protective Groups in Organic Synthesis, 3rd ed.; Wiley: New York, 1999.
- [12] Lay, L.; Manzoni, L.; Schmidt, R. R. Carbohydr Res 1998, 310, 157–171.
- [13] Wang, P.; Shen, G.-J.; Wang, Y.-F.; Ichikawa, Y.; Wong, C.-H. J Org Chem 1993, 58, 3985–3990.
- [14] Bernet, B.; Vasella, A. Helv Chim Acta 1979, 62, 2400– 2410.
- [15] For a review about glycosylation reactions see: (a) Hanessian, S. In Preparative Carbohydrate Chemisry; Hanessian, S. (Ed.); Marcel Dekker: New York, 1997; (b) Collins, P.; Ferrier, R. Monosaccharides: Their Chemistry and Their Roles in Natural Products; Wiley: New York, 1995.
- [16] Lafont, D.; Boullanger, P.; Chierici, S. New J Chem 1996, 20, 1093–1101.
- [17] (a) Wang, W.; Kong, F. J Org Chem 1998, 63, 5744– 5745; (b) Wang, W.; Kong, F. Tetrahedron Lett 1998, 39, 1937–1940; (c) Crich, D.; Dai, Z.; Gastaldi, S. J Org Chem 1999, 64, 5224–5229; (d) Wang, W.; Kong, F. Tetrahedron Lett 1999, 40, 1361–1364; (e) Wang, W.; Kong, F. J Org Chem 1999, 64, 5091–5095; (f) Wang, W.; Kong, F. Angew Chem Int Ed Engl 1999, 38, 1247–1250.